

Comprehensive additive release and bioaccessibility model for risk assessment of micro- and nano-plastics in the environment

P. Lee Ferguson and Mark R. Wiesner

Department of Civil & Environmental Engineering, Duke University, USA

Primary Objective: Create a robust and generalizable model to predict polymer additive release, transformation (where relevant), and bioaccessibility in context of realistic aquatic environments

This objective will be addressed through the following work packages (WP):

WP #1: Compile a comprehensive literature review of polymer additive chemical space, application by polymer type, transformation, and leachability in context of ambient waters and ecological receptor ingestion.

WP #2: Build and test a quantitative model to predict additive distribution among environmental compartments including polymer particles, water, and digestive environments representing deposit feeders, filter feeders, aquatic predators, and humans.

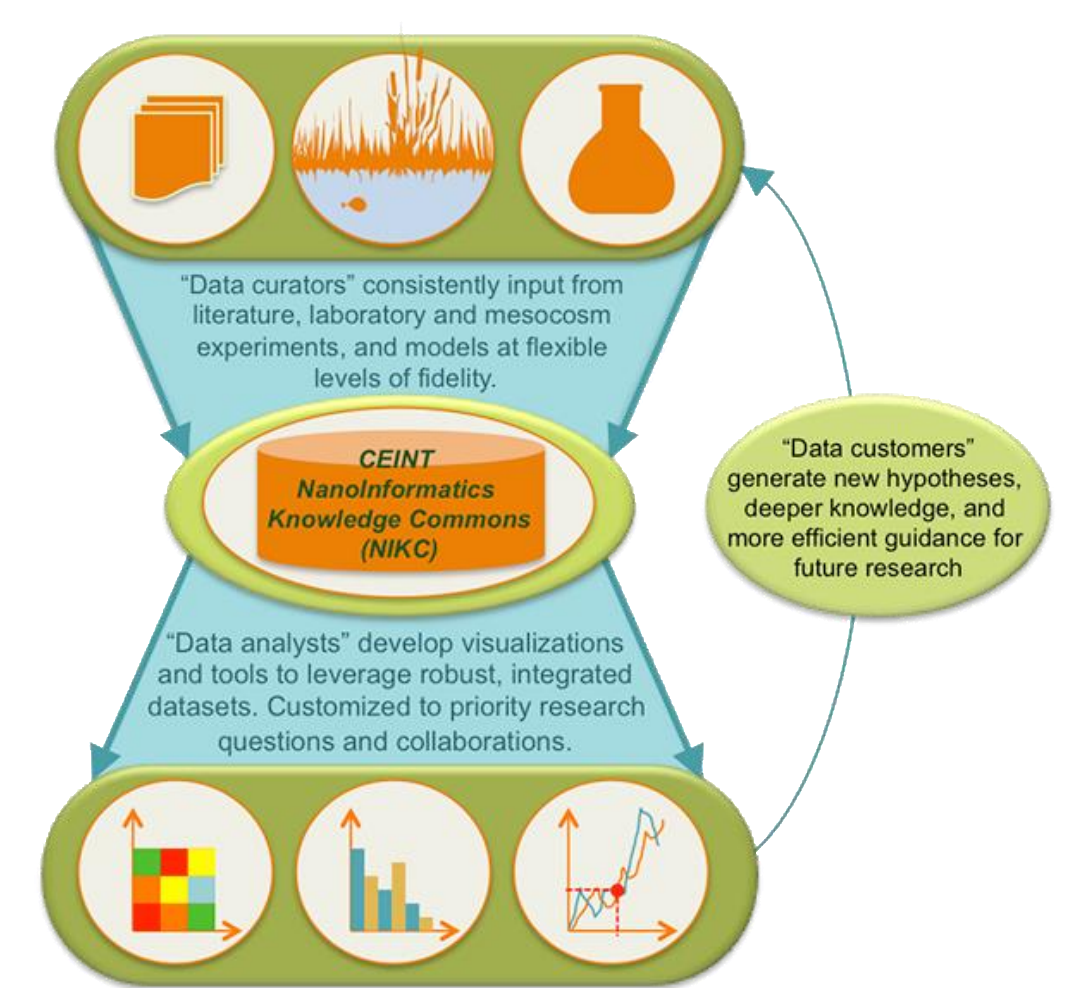
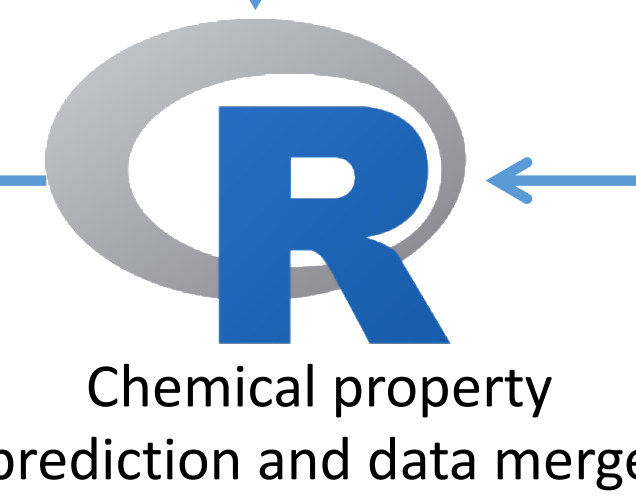
WP #3: Perform laboratory-based polymer additive leaching/bioaccessibility experiments designed to parameterize, validate, and test the model built in WP# 2.

Literature review and additive database

The primary output will be a three-element database designed to capture current knowledge associated with three data types:

- Data Type 1:** Additive properties, based on measured or predicted molecular/material parameters
- Data Type 2:** Polymer properties, based on bulk performance/manufacturing parameters of common-use and performance plastics
- Data Type 3:** Environmental/receptor properties, describing relevant parameters of natural waters and receptor organism gut physiology

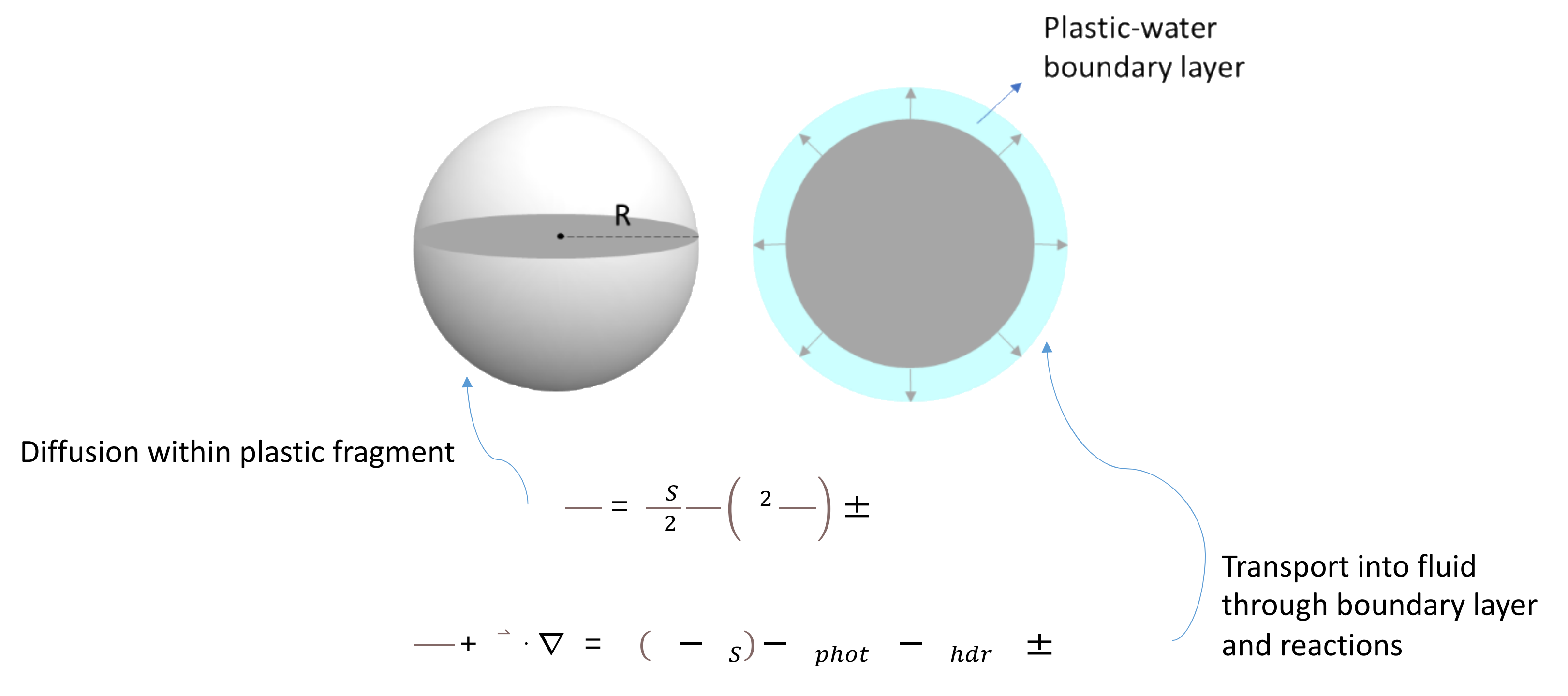
A	B	C	D	E	F	G
1	CAS	Name	Synonym	InChI	SMILES	Formula
2	100-20-9	terephthaloyl chloride	TREPHTHALOYL DICHLORIDE	ClC1=CC=C(C=C1)C(=O)Cl	ClC1=CC=C(C=C1)C(=O)Cl	C8H6Cl2O2
3	100-21-0	terephthalic acid	1,4-Benzenedicarboxylic acid	OC(=O)C1=CC=C(C=C1)C(=O)O	OC(=O)C1=CC=C(C=C1)C(=O)O	C8H6O4
4	100-37-8	2,2-BIS(4-METHYLPHENYL) ETHANE	2,2-DIPHENYLPROPANE	CC1=CC=C(C=C1)C(C)(C)C2=CC=CC=C2	CC1=CC=C(C=C1)C(C)(C)C2=CC=CC=C2	C15H14
5	100-41-4	Styrene	Ethylbenzene	C=CC1=CC=CC=C1	C=CC1=CC=CC=C1	C8H8
6	100-42-5	Styrene	Ethylbenzene	C=CC1=CC=CC=C1	C=CC1=CC=CC=C1	C8H8
7	100-44-7	benzyl chloride	alpha-chlorotoluene	ClCC1=CC=CC=C1	ClCC1=CC=CC=C1	C7H7Cl
8	100-51-6	benzyl alcohol	Phenylmethanol	OC1=CC=CC=C1CO	OC1=CC=CC=C1CO	C7H8O
9	100-52-7	benzyl acetate	Acetyl benzoate	CC(=O)OC1=CC=CC=C1	CC(=O)OC1=CC=CC=C1	C9H10O2
10	100-52-2	phenylacetylene	Acetylene	C#CC1=CC=CC=C1	C#CC1=CC=CC=C1	C8H6
11	100-74-3	4-ethylphenylamine	Amphetamine	CC1=CC=C(C=C1)CN	CC1=CC=C(C=C1)CN	C9H11N
12	100-78-9	benzene	benzene	C1=CC=CC=C1	C1=CC=CC=C1	C6H6
13	100-84-6	4-aminobenzonitrile	4-aminobenzonitrile	N#CC1=CC=C(N)C=C1	N#CC1=CC=C(N)C=C1	C7H6N2
14	100-97-0	benzoin	benzoin	O=C(OCC1=CC=CC=C1)O	O=C(OCC1=CC=CC=C1)O	C9H8O3
15	100-99-2	nitrobenzene	nitrobenzene	[O-][N+](=O)C1=CC=CC=C1	[O-][N+](=O)C1=CC=CC=C1	C6H5NO2
16	100-98-7	nitrobenzene	nitrobenzene	[O-][N+](=O)C1=CC=CC=C1	[O-][N+](=O)C1=CC=CC=C1	C6H5NO2
17	100-201-2	alpha-bromoacetone	alpha-bromoacetone	BrCC(=O)C	BrCC(=O)C	C2H3BrO
18	100-53-5	di-n-butylamine	di-n-butylamine	CCCCN(C)CCCC	CCCCN(C)CCCC	C12H27N
19	100-86-2	peroxyacetic acid	peroxyacetic acid	CC(=O)OC(=O)O	CC(=O)OC(=O)O	C2H4O3
20	100-22-1	barium nitrate	barium nitrate	[O-][N+](=O)[O-].[Ba+2]	[O-][N+](=O)[O-].[Ba+2]	BaO4N2
21	100-58-1	1,2-ethanediol	ethylene glycol	OCCO	OCCO	C2H6O2
22	100-25-9	calcium chloride	calcium chloride	[Cl-].[Ca+2]	[Cl-].[Ca+2]	CaCl2



Additive release model development and testing

The primary output will be numerical code that can be integrated as a subroutine in a larger particle- population model for particulate plastics.

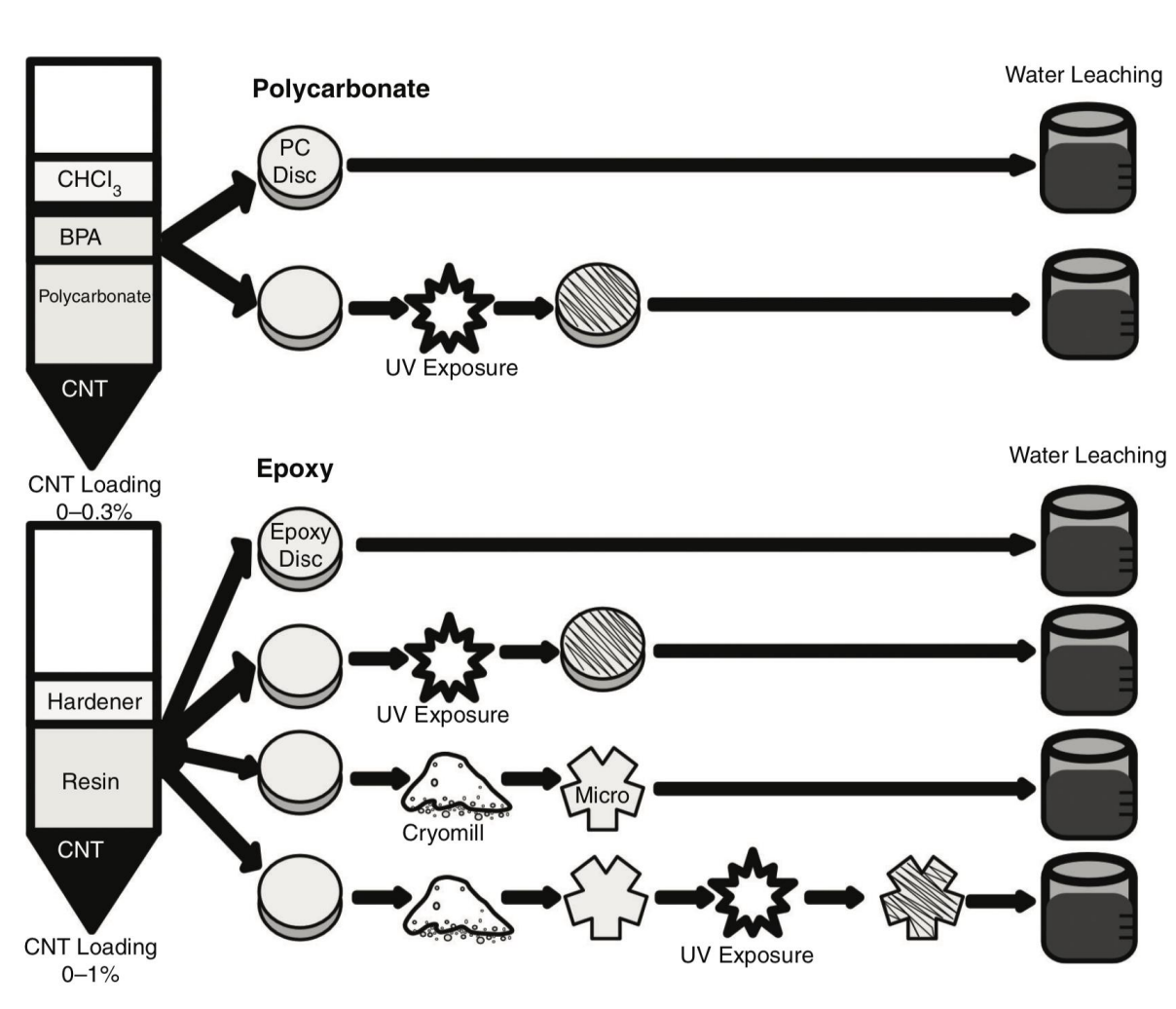
- Mathematical formulation for several cases:
 - Non-reactive additives
 - Reactive additives and pertinent degradation reactions
 - Non-homogeneous distribution within fragments
- Calculation of additive release in a variety of media including:
 - Ocean water
 - Fresh water
 - Representative vertebrate gut environments



Additive leaching and bioaccessibility experiments

Experiments will be designed to explore:

- Influence of additive chemical structure on leachability (QSAR approach)
- Polymer type/geometry/chemistry as a driving factor for leachability
- Impact of type and extent of weathering (e.g. physical vs. chemical vs. photolytic) on additive leaching
- Additive localization on/in polymer and its influence on leaching
- Gut digestive fluid parameters and their importance to additive leaching from polymers
- Reactivity and transformation of leachable additives



Adapted from: Walker, I. et al., *Environ. Chem.* 2021, 18, 131–141

Additive Type	<ul style="list-style-type: none"> Surface-coating (e.g. PFAS components for stain resistance/water-repellency) Intrinsic additive (e.g. disperse azobenzene dyes in polyester microfibers) Performance additive (e.g. antioxidants such as Irgafos 168 or UV inhibitors such as Tinuvin 770)
Polymer Type	<ul style="list-style-type: none"> Polyolefin (relatively inert to direct UV degradation, e.g. polypropylene) Aromatic thermoplastic (glassy and UV-active, e.g. polystyrene) Elastomer (rubbery and subject to oxidation, e.g. polyurethane)
Weathering Treatment	<ul style="list-style-type: none"> UV exposure (pre-treatment of polymer with SUNTEST XLS+ in laboratory) Physical abrasion/fragmentation (laboratory abradant or cryomill) Oxidant exposure (pre-treatment with ozone or peroxide)
Leaching Treatment	<ul style="list-style-type: none"> Water (fresh-to-brackish, pH 5-8, dissolved organic matter, temperature 4° - 40° C) Simulated or actual digestive fluid (bile salt surfactants, lipase, pepsin, other enzymes, pH 2-7)